HIGHFIELDS SCHOOL

CURRICULUM OVERVIEW 2023-2024



SUBJECT: A LEVEL CHEMISTRY EXAMINATION BOARD: OCR		
AUTUMN TERM - YEAR 12	SPRING TERM - YEAR 12	SUMMER TERM - YEAR 12
 H032 Module 1 - Development of practical skills Module 2 - Foundation in Chemistry 2.1 - Atomic structure, isotopes, formulae, equations, moles, acids, redox 2.2 - Electron structure, bonding and structure Module 3 - Periodic table and Energy 3.1 - Periodicity, groups 2 and 7, qualitative analysis 3.2 - Enthalpy changes reaction rates, equilibrium Module 4 - Core Organic Chemistry 4.1 - Basic concepts, alkanes, alkenes Potential Practical Activity Moles determination Acid-alkali titrations Reactions involving Group 2 and Group 7 Qualitative analysis of ions 	 Module 1 - Development of practical skills Module 4 - Core Organic Chemistry continued 4.1 - Basic concepts, alkanes, alkenes 4.2 - Alcohols, haloalkanes, organic synthesis, analytical synthesis (IR, MS) Revision and Exam preparation (4Rs) Mini-tests and quizzes Past examination questions Structured and multiple-choice questions Potential Practical Activity Measuring enthalpy change Alkane/Alkene reactions Oxidation reactions Preparing haloalkanes Synthesis of esters 	 H432 Module 1 - Development of practical skills Module 5 - Physical Chemistry, Transition elements 5.1 - How far, how fast 5.3 - Transition elements Module 6 - Organic Chemistry and analysis 6.1 - Aromatic compounds, carbonyls Potential Practical Activity Rates: continuous monitoring and initial rates pH measurements lodine clock; Thiosulfate clock Acid-base titrations Redox titrations Transition metal precipitation and ligand substitution
 ASSESSMENT Transition GCSE exams C1-C3 and C4-C6. 2.1.1 Atomic structure and isotopes. 2.1.3/2.1.4 amount, acids and bases. 2.1.5 Redox. 2.2.1/2.2.2 electron, bonding and structure. 4.1 Basic concepts and hydrocarbons. 	 ASSESSMENT 3.1 Periodic table. 3.2.1 Enthalpy changes. 4.2 Alcohols, Halogenoalkanes and analysis. 3.2.2/3.2.3 reaction rates and equilibrium. Practice exam: H032 Paper 1 (50%, 70 marks, 90mins) 	ASSESSMENT • 5.3.1 Transition elements. • 6.1.1 Aromatic compounds. School exams H032 Paper 1 (50%, 70 marks, 90mins) H032 Paper 2 (50%, 70 marks, 90mins) Re-sit exam HO32/01 paper 1 (50%, 70 marks, 90mins)

HIGHFIELDS SCHOOL

CURRICULUM OVERVIEW 2023-2024



SUBJECT: A LEVEL CHEMISTRY EXAMINATION BOARD: OCR		
AUTUMN TERM - YEAR 13	SPRING TERM - YEAR 13	SUMMER TERM - YEAR 13
H432 Module 1 - Development of practical skills Module 5 - Physical Chemistry, Transition elements • 5.1 - How far, how fast, acid bases and buffers	 Module 1 - Development of practical skills Module 5 - Physical Chemistry, Transition elements 5.2 - Lattice enthalpy, enthalpy and entropy, redox, electrode potentials 	 Revision and Exam preparation (4Rs) Mini-tests and Seneca quizzes Past examination question booklets Structured and multiple-choice questions H432 past exam papers
 5.2 - Lattice enthalpy, enthalpy and entropy, redox, electrode potentials 5.3 - Transition elements, qualitative analysis Module 6 - Organic Chemistry and analysis 6.1 - Aromatic compounds, carbonyls, carboxylic acids and esters 	 Module 6 - Organic Chemistry and analysis 6.2 - Amines, amino acids, amides and chirality, polyesters and polyamides, carbon-carbon bond formation, organic synthesis 6.3 - Chromatography, qualitative analysis, spectroscopy 	
 6.2 - Amines, amino acids, amides and chirality, polyesters and polyamides, carbon-carbon bond formation, organic synthesis 6.3 - Chromatography, qualitative analysis, spectroscopy 	 Revision and Exam preparation (4Rs) Mini-tests and quizzes for all modules Past examination questions Structured and multiple-choice questions H032 paper 1 and paper 2 resources 	
 Potential Practical Activities Rates: continuous monitoring and initial rates pH measurements lodine clock; Thiosulfate clock Acid-base titrations Measuring enthalpy change Electrochemical cells 	 Potential Practical Activities Transition metal precipitation and ligand substitution Redox titrations Qualitative analysis: identifying functional groups Synthesis of organic solids Preparing 2,4-DNPH derivatives Synthesis of Paracetamol 	
 ASSESSMENT 5.1 rates equilibrium and pH. 5.3 Transition elements. 6.1 Aromatic compounds, carbonyls, acids. Practice exam: H032 Paper 1 	 ASSESSMENT 5.2 Energy 6.2 Nitrogen compounds, polymers, synthesis 6.3 Analysis School exams: H432 Paper 1 and Paper 2 Practice exams: H432 Paper 1, Paper 2, Paper 3 	ASSESSMENT • Practice exams: H432 Paper 1, Paper 2 and Paper 3 Final examinations H432/01 Paper 1 (37%, 100 marks, 135mins) H432/02 Paper 2 (37%, 100 marks, 135mins) H432/03 Paper 3 (26%, 70 marks, 90mins)